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2-(4-Chlorophenyl)-2-oxoethyl 2,4-difluorobenzoate

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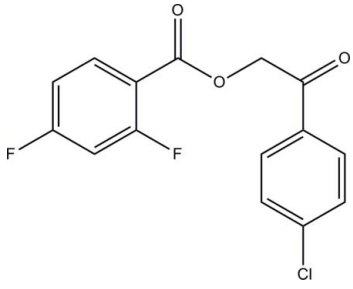
Key indicators: single-crystal X-ray study; T = 296 K; mean σ(C–C) = 0.005 Å; R factor = 0.054; wR factor = 0.206; data-to-parameter ratio = 16.6.

The asymmetric unit of title compound, $C_{15}H_9ClF_2O_3$, consists of two crystallographically independent molecules. The dihedral angle between the two terminal benzene rings in one molecule is 7.92 (14)°, while that in the other molecule is 73.50 (16)°. In the crystal, molecules are stacked into columns along the b axis by intermolecular C—H···O hydrogen bonds. A π–π interaction with a centroid-to-centroid distance of 3.747 (2) Å further stabilizes the crystal structure.

Related literature

For background to and applications of phenacyl benzoates, see: Rather & Reid (1919); Sheehan & Umezaw (1973); Ruzicka *et al.* (2002); Litera *et al.* (2006); Huang *et al.* (1996); Gandhi *et al.* (1995). For reference bond-length values, see: Allen *et al.* (1987).



Experimental

Crystal data

$C_{15}H_9ClF_2O_3$
M_r = 310.67
Monoclinic, $P2_1/c$
a = 16.0179 (17) Å
b = 7.9609 (8) Å
c = 24.0172 (18) Å
β = 115.939 (5)°
V = 2754.1 (5) Å^3
Z = 8
Mo $K\alpha$ radiation
μ = 0.31 mm^{-1}
T = 296 K
0.55 × 0.26 × 0.09 mm

Data collection

Bruker SMART APEXII DUO CCD area-detector diffractometer
Absorption correction: multi-scan (*SADABS*; Bruker, 2009)
T_{min} = 0.850, T_{max} = 0.974
17424 measured reflections
6308 independent reflections
3353 reflections with $I > 2\sigma(I)$
R_{int} = 0.030

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.054
$wR(F^2)$ = 0.206
S = 1.03
6308 reflections
379 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.33 e Å^{-3}
$\Delta\rho_{min}$ = −0.32 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
C8A—H8AB···O3B	0.97	2.60	3.451 (4)	147
C8A—H8AA···O1B^{i}	0.97	2.42	3.294 (3)	149
C5B—H5BA···O3A^{ii}	0.93	2.50	3.376 (4)	158
C8B—H8BB···O3A^{ii}	0.97	2.58	3.415 (3)	144
C14B—H14B···O1A^{iii}	0.93	2.59	3.216 (5)	125

Symmetry codes: (i) $x, y-1, z$; (ii) $-x+1, y+\frac{1}{2}, -z+\frac{3}{2}$; (iii) $-x+1, -y+1, -z+2$.

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: IS2722).

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supplementary materials

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2-(4-Chlorophenyl)-2-oxoethyl 2,4-difluorobenzoate

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Comment

Phenacyl benzoates derivatives are very important in identification of organic acids (Rather & Reid, 1919), they undergo photolysis in neutral and mild conditions (Sheehan & Umezaw, 1973; Ruzicka *et al.*, 2002; Litera *et al.*, 2006). They find applications in the field of synthetic chemistry for the synthesis of oxazoles, imidazoles (Huang *et al.*, 1996), benzoxazepine (Gandhi *et al.*, 1995). We hereby report the crystal structure of 2-(4-chlorophenyl)-2-oxoethyl 2,4-difluorobenzoate of potential commercial importance.

The asymmetric unit of the title compound (Fig. 1), consists of two crystallographically independent molecules *A* and *B*. Both terminal phenyl rings (C1–C6 and C10–C15) in molecules *A* and *B* make dihedral angles of 7.92 (14) and 73.50 (16)° to each other, respectively. The bond lengths (Allen *et al.*, 1987) and angles are within normal ranges.

The crystal packing is shown in Fig. 2. The intermolecular C8A—H8AB···O3B hydrogen bond linked the molecule *A* with molecule *B* together. The molecules are linked into columns along the *b* axis by the intermolecular C8A—H8AA···O1B, C5B—H5BA···O3A, C8B—H8BB···O3A and C14B—H14B···O1A hydrogen bonds (Table 1). A π–π interaction further stabilizes the crystal structure [$Cg1\cdots Cg2^{ii}$ = 3.747 (2) Å; *Cg*1 and *Cg*2 are centroids of C1B–C6B and C10B–C15B benzene ring, respectively].

Experimental

A mixture of 2,4-difluorobenzoic acid (1.0 g, 0.0063 mol) potassium carbonate (0.95 g, 0.0069 mol) and 2-bromo-1-(4-chlorophenyl)ethanone (1.41 g, 0.0063 mol) in dimethylformamide (10 ml) was stirred at room temperature for 2 h. On cooling, colourless needle-shaped crystals 2-(4-chlorophenyl)-2-oxoethyl 2,4-difluorobenzoate begin to separate. It was collected by filtration and recrystallized from ethanol. Yield: 1.65 g, 84.1%. M.p.: 376–377 K.

Refinement

All H atoms were positioned geometrically (C—H = 0.93 or 0.97 Å) and refined using a riding model with U_{iso}(H) = 1.2U_{eq}(C). Four reflections, -6 4 4, -6 5 8, -1 1 1 and -1 4 10, were omitted.

Figures

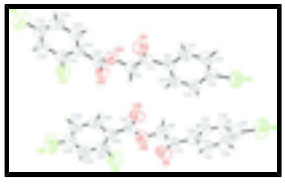

Fig. 1. The molecular structure of the title compound, showing two independent molecules with atom labels with 50% probability displacement ellipsoids.

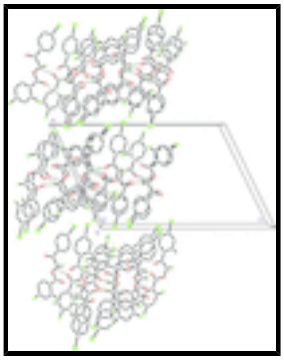

Fig. 2. The crystal packing of the title compound. Dashed lines represent the hydrogen bonds.

2-(4-Chlorophenyl)-2-oxoethyl 2,4-difluorobenzoate

Crystal data

$C_{15}H_9ClF_2O_3$	$F(000) = 1264$
$M_r = 310.67$	$D_x = 1.499$ Mg m^{-3}
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Hall symbol: -P 2ybc	Cell parameters from 3882 reflections
$a = 16.0179$ (17) Å	$\theta = 2.6$–$22.5°$
$b = 7.9609$ (8) Å	$\mu = 0.31$ mm^{-1}
$c = 24.0172$ (18) Å	$T = 296$ K
$\beta = 115.939$ (5)°	Needle, colourless
$V = 2754.1$ (5) Å^3	0.55 × 0.26 × 0.09 mm
$Z = 8$	

Data collection

Bruker SMART APEXII DUO CCD area-detector diffractometer	6308 independent reflections
Radiation source: fine-focus sealed tube	3353 reflections with $I > 2\sigma(I)$
graphite	$R_{int} = 0.030$
φ and ω scans	$\theta_{max} = 27.5°$, $\theta_{min} = 1.8°$
Absorption correction: multi-scan (*SADABS*; Bruker, 2009)	$h = -20\rightarrow19$
$T_{min} = 0.850$, $T_{max} = 0.974$	$k = -10\rightarrow10$
17424 measured reflections	$l = -31\rightarrow31$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.054$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.206$	H-atom parameters constrained
$S = 1.03$	$w = 1/[\sigma^2(F_o^2) + (0.114P)^2 + 0.0306P]$ where $P = (F_o^2 + 2F_c^2)/3$

6308 reflections	$(\Delta/\sigma)_{max} = 0.001$
379 parameters	$\Delta\rho_{max} = 0.33$ e Å^{-3}
0 restraints	$\Delta\rho_{min} = -0.32$ e Å^{-3}

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
Cl1A	0.01833 (6)	0.69373 (15)	0.67059 (4)	0.1116 (4)
F1A	0.77697 (12)	0.3081 (3)	1.00087 (8)	0.1006 (6)
F2A	0.80582 (16)	−0.0433 (3)	1.16292 (10)	0.1139 (7)
O1A	0.32955 (14)	0.3728 (3)	0.93888 (9)	0.0827 (6)
O2A	0.49808 (12)	0.3539 (2)	0.95029 (8)	0.0654 (5)
O3A	0.60473 (13)	0.3792 (3)	0.91508 (8)	0.0869 (7)
C1A	0.1715 (2)	0.4728 (4)	0.83169 (14)	0.0734 (8)
H1AA	0.1615	0.4076	0.8603	0.088*
C2A	0.0965 (2)	0.5297 (4)	0.77968 (15)	0.0828 (9)
H2AA	0.0364	0.5010	0.7727	0.099*
C3A	0.11135 (19)	0.6286 (4)	0.73861 (13)	0.0718 (7)
C4A	0.1996 (2)	0.6748 (4)	0.74836 (13)	0.0710 (7)
H4AA	0.2086	0.7461	0.7208	0.085*
C5A	0.27448 (19)	0.6134 (4)	0.79980 (12)	0.0661 (7)
H5AA	0.3343	0.6413	0.8061	0.079*
C6A	0.26202 (17)	0.5109 (3)	0.84226 (11)	0.0570 (6)
C7A	0.34074 (18)	0.4379 (3)	0.89723 (11)	0.0589 (6)
C8A	0.43456 (17)	0.4445 (3)	0.89777 (11)	0.0608 (6)
H8AA	0.4319	0.3949	0.8601	0.073*
H8AB	0.4547	0.5603	0.8999	0.073*
C9A	0.58203 (18)	0.3275 (3)	0.95319 (11)	0.0590 (6)
C10A	0.64078 (17)	0.2283 (3)	1.00863 (11)	0.0573 (6)
C11A	0.73636 (19)	0.2215 (4)	1.03031 (12)	0.0667 (7)
C12A	0.7933 (2)	0.1312 (4)	1.08180 (13)	0.0754 (8)
H12A	0.8572	0.1285	1.0954	0.091*
C13A	0.7515 (2)	0.0455 (4)	1.11226 (13)	0.0760 (8)
C14A	0.6584 (2)	0.0462 (4)	1.09386 (13)	0.0758 (8)
H14A	0.6325	−0.0152	1.1154	0.091*
C15A	0.6027 (2)	0.1400 (3)	1.04240 (12)	0.0649 (7)

H15A	0.5390	0.1444	1.0301	0.078*
Cl1B	−0.01753 (7)	1.09979 (18)	0.57396 (5)	0.1307 (5)
F1B	0.70241 (15)	1.0849 (3)	0.83037 (10)	0.1280 (9)
F2B	0.92324 (16)	0.8868 (4)	1.01835 (10)	0.1338 (8)
O1B	0.41561 (18)	1.1476 (3)	0.79738 (10)	0.1000 (7)
O2B	0.54117 (14)	0.9566 (3)	0.78473 (8)	0.0834 (6)
O3B	0.49789 (16)	0.8028 (3)	0.84487 (11)	0.1036 (8)
C1B	0.2248 (3)	1.1648 (4)	0.71976 (14)	0.0828 (9)
H1BA	0.2467	1.2199	0.7577	0.099*
C2B	0.1322 (3)	1.1733 (5)	0.67968 (16)	0.0935 (10)
H2BA	0.0915	1.2341	0.6902	0.112*
C3B	0.1000 (2)	1.0906 (4)	0.62378 (13)	0.0838 (9)
C4B	0.1590 (2)	1.0031 (4)	0.60703 (12)	0.0816 (9)
H4BA	0.1365	0.9495	0.5688	0.098*
C5B	0.2515 (2)	0.9953 (4)	0.64709 (11)	0.0717 (7)
H5BA	0.2916	0.9357	0.6358	0.086*
C6B	0.2865 (2)	1.0750 (3)	0.70449 (11)	0.0644 (7)
C7B	0.3848 (2)	1.0667 (3)	0.74964 (12)	0.0696 (7)
C8B	0.4479 (2)	0.9489 (5)	0.73696 (12)	0.0805 (9)
H8BA	0.4248	0.8350	0.7339	0.097*
H8BB	0.4475	0.9778	0.6976	0.097*
C9B	0.5572 (2)	0.8795 (4)	0.83760 (13)	0.0702 (7)
C10B	0.65445 (19)	0.8922 (3)	0.88491 (12)	0.0629 (7)
C11B	0.7239 (2)	0.9844 (4)	0.88009 (13)	0.0757 (8)
C12B	0.8133 (2)	0.9838 (5)	0.92315 (15)	0.0882 (9)
H12B	0.8582	1.0466	0.9179	0.106*
C13B	0.8352 (2)	0.8877 (5)	0.97465 (14)	0.0862 (9)
C14B	0.7698 (3)	0.7944 (4)	0.98378 (14)	0.0864 (9)
H14B	0.7858	0.7300	1.0193	0.104*
C15B	0.6810 (2)	0.8001 (4)	0.93878 (13)	0.0757 (8)
H15B	0.6359	0.7390	0.9445	0.091*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cl1A	0.0784 (6)	0.1292 (9)	0.1112 (7)	0.0192 (5)	0.0266 (5)	0.0172 (6)
F1A	0.0685 (11)	0.1358 (18)	0.1080 (13)	−0.0061 (10)	0.0484 (10)	0.0250 (11)
F2A	0.1271 (17)	0.1155 (17)	0.1067 (13)	0.0402 (13)	0.0580 (13)	0.0393 (12)
O1A	0.0813 (13)	0.0993 (16)	0.0837 (12)	−0.0051 (11)	0.0511 (11)	0.0162 (11)
O2A	0.0666 (11)	0.0714 (12)	0.0692 (10)	0.0014 (9)	0.0398 (9)	0.0027 (9)
O3A	0.0691 (12)	0.130 (2)	0.0746 (11)	−0.0026 (11)	0.0432 (10)	0.0174 (11)
C1A	0.0694 (18)	0.0748 (19)	0.0922 (18)	−0.0099 (15)	0.0504 (16)	0.0013 (15)
C2A	0.0612 (17)	0.091 (2)	0.107 (2)	−0.0066 (15)	0.0467 (17)	0.0037 (18)
C3A	0.0655 (17)	0.0686 (18)	0.0827 (17)	0.0034 (14)	0.0337 (14)	−0.0080 (14)
C4A	0.0788 (19)	0.0675 (18)	0.0772 (16)	0.0013 (14)	0.0437 (15)	0.0051 (14)
C5A	0.0656 (16)	0.0652 (17)	0.0790 (16)	−0.0064 (13)	0.0420 (14)	−0.0023 (13)
C6A	0.0631 (15)	0.0524 (14)	0.0665 (13)	−0.0075 (11)	0.0386 (12)	−0.0110 (11)
C7A	0.0722 (17)	0.0505 (14)	0.0670 (14)	−0.0098 (12)	0.0426 (13)	−0.0107 (12)

C8A	0.0652 (16)	0.0622 (16)	0.0629 (13)	−0.0041 (12)	0.0352 (12)	−0.0022 (12)
C9A	0.0619 (15)	0.0620 (16)	0.0619 (13)	−0.0168 (12)	0.0352 (12)	−0.0134 (12)
C10A	0.0647 (15)	0.0523 (15)	0.0632 (13)	−0.0099 (12)	0.0356 (12)	−0.0121 (11)
C11A	0.0684 (17)	0.0688 (18)	0.0739 (15)	−0.0073 (14)	0.0414 (14)	−0.0037 (13)
C12A	0.0723 (18)	0.078 (2)	0.0825 (17)	0.0069 (15)	0.0396 (15)	−0.0008 (15)
C13A	0.096 (2)	0.0638 (18)	0.0739 (16)	0.0151 (16)	0.0419 (16)	0.0045 (14)
C14A	0.099 (2)	0.0632 (18)	0.0848 (18)	−0.0055 (16)	0.0580 (18)	0.0019 (15)
C15A	0.0722 (17)	0.0585 (16)	0.0745 (15)	−0.0096 (13)	0.0416 (14)	−0.0075 (13)
Cl1B	0.0942 (7)	0.1856 (13)	0.1040 (7)	0.0478 (7)	0.0356 (6)	0.0041 (7)
F1B	0.1063 (16)	0.154 (2)	0.1211 (15)	−0.0321 (14)	0.0474 (13)	0.0536 (14)
F2B	0.0901 (15)	0.181 (2)	0.1081 (15)	−0.0141 (15)	0.0231 (13)	0.0130 (15)
O1B	0.1165 (19)	0.0871 (16)	0.0938 (15)	−0.0155 (13)	0.0435 (14)	−0.0334 (12)
O2B	0.0794 (14)	0.1104 (17)	0.0709 (11)	−0.0151 (12)	0.0426 (11)	−0.0034 (11)
O3B	0.0879 (16)	0.0978 (17)	0.1253 (18)	−0.0294 (13)	0.0468 (14)	0.0205 (14)
C1B	0.109 (3)	0.075 (2)	0.0801 (18)	0.0001 (17)	0.0553 (19)	−0.0151 (15)
C2B	0.106 (3)	0.095 (3)	0.096 (2)	0.026 (2)	0.059 (2)	−0.0040 (19)
C3B	0.085 (2)	0.097 (2)	0.0764 (17)	0.0228 (17)	0.0423 (16)	0.0098 (16)
C4B	0.087 (2)	0.102 (2)	0.0608 (15)	0.0152 (17)	0.0365 (16)	−0.0038 (15)
C5B	0.085 (2)	0.078 (2)	0.0655 (15)	0.0082 (15)	0.0451 (15)	−0.0012 (13)
C6B	0.0864 (19)	0.0554 (16)	0.0664 (14)	−0.0021 (13)	0.0472 (14)	0.0018 (12)
C7B	0.093 (2)	0.0588 (17)	0.0702 (15)	−0.0190 (14)	0.0481 (16)	−0.0099 (13)
C8B	0.077 (2)	0.103 (2)	0.0665 (15)	−0.0066 (16)	0.0357 (15)	−0.0138 (15)
C9B	0.0790 (19)	0.0623 (17)	0.0821 (17)	−0.0141 (15)	0.0470 (15)	−0.0104 (14)
C10B	0.0774 (18)	0.0541 (16)	0.0727 (15)	−0.0107 (13)	0.0473 (14)	−0.0084 (12)
C11B	0.089 (2)	0.0713 (19)	0.0778 (17)	−0.0086 (16)	0.0466 (17)	0.0091 (15)
C12B	0.080 (2)	0.100 (3)	0.094 (2)	−0.0243 (18)	0.0477 (18)	−0.0006 (19)
C13B	0.075 (2)	0.103 (3)	0.0771 (18)	−0.0093 (18)	0.0295 (16)	−0.0089 (17)
C14B	0.107 (3)	0.086 (2)	0.0711 (17)	−0.0115 (19)	0.0433 (18)	0.0055 (16)
C15B	0.095 (2)	0.0682 (19)	0.0792 (17)	−0.0194 (15)	0.0523 (17)	−0.0038 (14)

Geometric parameters (Å, °)

Cl1A—C3A	1.740 (3)	Cl1B—C3B	1.735 (3)
F1A—C11A	1.342 (3)	F1B—C11B	1.351 (3)
F2A—C13A	1.346 (3)	F2B—C13B	1.340 (4)
O1A—C7A	1.207 (3)	O1B—C7B	1.216 (3)
O2A—C9A	1.332 (3)	O2B—C9B	1.330 (3)
O2A—C8A	1.423 (3)	O2B—C8B	1.432 (3)
O3A—C9A	1.196 (3)	O3B—C9B	1.204 (3)
C1A—C2A	1.377 (4)	C1B—C2B	1.372 (5)
C1A—C6A	1.392 (4)	C1B—C6B	1.393 (4)
C1A—H1AA	0.9300	C1B—H1BA	0.9300
C2A—C3A	1.362 (4)	C2B—C3B	1.377 (5)
C2A—H2AA	0.9300	C2B—H2BA	0.9300
C3A—C4A	1.378 (4)	C3B—C4B	1.368 (4)
C4A—C5A	1.381 (4)	C4B—C5B	1.370 (4)
C4A—H4AA	0.9300	C4B—H4BA	0.9300
C5A—C6A	1.387 (3)	C5B—C6B	1.393 (3)
C5A—H5AA	0.9300	C5B—H5BA	0.9300

C6A—C7A	1.488 (4)	C6B—C7B	1.472 (4)
C7A—C8A	1.498 (3)	C7B—C8B	1.504 (4)
C8A—H8AA	0.9700	C8B—H8BA	0.9700
C8A—H8AB	0.9700	C8B—H8BB	0.9700
C9A—C10A	1.479 (4)	C9B—C10B	1.477 (4)
C10A—C11A	1.386 (4)	C10B—C11B	1.380 (4)
C10A—C15A	1.399 (3)	C10B—C15B	1.382 (4)
C11A—C12A	1.375 (4)	C11B—C12B	1.350 (4)
C12A—C13A	1.370 (4)	C12B—C13B	1.363 (4)
C12A—H12A	0.9300	C12B—H12B	0.9300
C13A—C14A	1.357 (4)	C13B—C14B	1.378 (5)
C14A—C15A	1.384 (4)	C14B—C15B	1.360 (4)
C14A—H14A	0.9300	C14B—H14B	0.9300
C15A—H15A	0.9300	C15B—H15B	0.9300
C9A—O2A—C8A	116.13 (18)	C9B—O2B—C8B	116.3 (2)
C2A—C1A—C6A	121.3 (3)	C2B—C1B—C6B	121.0 (3)
C2A—C1A—H1AA	119.3	C2B—C1B—H1BA	119.5
C6A—C1A—H1AA	119.3	C6B—C1B—H1BA	119.5
C3A—C2A—C1A	119.2 (3)	C1B—C2B—C3B	119.3 (3)
C3A—C2A—H2AA	120.4	C1B—C2B—H2BA	120.4
C1A—C2A—H2AA	120.4	C3B—C2B—H2BA	120.4
C2A—C3A—C4A	121.5 (3)	C4B—C3B—C2B	121.2 (3)
C2A—C3A—Cl1A	120.2 (2)	C4B—C3B—Cl1B	120.0 (2)
C4A—C3A—Cl1A	118.3 (2)	C2B—C3B—Cl1B	118.8 (2)
C3A—C4A—C5A	118.8 (3)	C3B—C4B—C5B	119.4 (3)
C3A—C4A—H4AA	120.6	C3B—C4B—H4BA	120.3
C5A—C4A—H4AA	120.6	C5B—C4B—H4BA	120.3
C4A—C5A—C6A	121.2 (2)	C4B—C5B—C6B	121.2 (3)
C4A—C5A—H5AA	119.4	C4B—C5B—H5BA	119.4
C6A—C5A—H5AA	119.4	C6B—C5B—H5BA	119.4
C5A—C6A—C1A	117.9 (3)	C1B—C6B—C5B	118.0 (3)
C5A—C6A—C7A	122.9 (2)	C1B—C6B—C7B	118.9 (2)
C1A—C6A—C7A	119.2 (2)	C5B—C6B—C7B	123.1 (2)
O1A—C7A—C6A	121.8 (2)	O1B—C7B—C6B	122.2 (3)
O1A—C7A—C8A	121.3 (2)	O1B—C7B—C8B	119.3 (3)
C6A—C7A—C8A	116.85 (19)	C6B—C7B—C8B	118.4 (2)
O2A—C8A—C7A	108.40 (19)	O2B—C8B—C7B	111.6 (2)
O2A—C8A—H8AA	110.0	O2B—C8B—H8BA	109.3
C7A—C8A—H8AA	110.0	C7B—C8B—H8BA	109.3
O2A—C8A—H8AB	110.0	O2B—C8B—H8BB	109.3
C7A—C8A—H8AB	110.0	C7B—C8B—H8BB	109.3
H8AA—C8A—H8AB	108.4	H8BA—C8B—H8BB	108.0
O3A—C9A—O2A	122.9 (2)	O3B—C9B—O2B	122.5 (3)
O3A—C9A—C10A	125.9 (2)	O3B—C9B—C10B	123.8 (3)
O2A—C9A—C10A	111.19 (19)	O2B—C9B—C10B	113.6 (2)
C11A—C10A—C15A	116.9 (2)	C11B—C10B—C15B	115.6 (3)
C11A—C10A—C9A	121.5 (2)	C11B—C10B—C9B	126.2 (2)
C15A—C10A—C9A	121.5 (2)	C15B—C10B—C9B	118.2 (2)
F1A—C11A—C12A	117.3 (2)	C12B—C11B—F1B	117.0 (3)

F1A—C11A—C10A	119.5 (2)	C12B—C11B—C10B	123.9 (3)
C12A—C11A—C10A	123.2 (2)	F1B—C11B—C10B	119.1 (3)
C13A—C12A—C11A	117.0 (3)	C11B—C12B—C13B	117.6 (3)
C13A—C12A—H12A	121.5	C11B—C12B—H12B	121.2
C11A—C12A—H12A	121.5	C13B—C12B—H12B	121.2
F2A—C13A—C14A	118.7 (3)	F2B—C13B—C12B	118.6 (3)
F2A—C13A—C12A	118.1 (3)	F2B—C13B—C14B	119.0 (3)
C14A—C13A—C12A	123.2 (3)	C12B—C13B—C14B	122.3 (3)
C13A—C14A—C15A	118.7 (3)	C15B—C14B—C13B	117.3 (3)
C13A—C14A—H14A	120.7	C15B—C14B—H14B	121.3
C15A—C14A—H14A	120.7	C13B—C14B—H14B	121.3
C14A—C15A—C10A	121.0 (3)	C14B—C15B—C10B	123.3 (3)
C14A—C15A—H15A	119.5	C14B—C15B—H15B	118.4
C10A—C15A—H15A	119.5	C10B—C15B—H15B	118.4
C6A—C1A—C2A—C3A	−1.6 (5)	C6B—C1B—C2B—C3B	0.4 (5)
C1A—C2A—C3A—C4A	−0.9 (5)	C1B—C2B—C3B—C4B	−1.3 (5)
C1A—C2A—C3A—Cl1A	176.8 (2)	C1B—C2B—C3B—Cl1B	179.1 (3)
C2A—C3A—C4A—C5A	2.6 (4)	C2B—C3B—C4B—C5B	1.1 (5)
Cl1A—C3A—C4A—C5A	−175.1 (2)	Cl1B—C3B—C4B—C5B	−179.3 (3)
C3A—C4A—C5A—C6A	−1.8 (4)	C3B—C4B—C5B—C6B	−0.1 (5)
C4A—C5A—C6A—C1A	−0.5 (4)	C2B—C1B—C6B—C5B	0.5 (4)
C4A—C5A—C6A—C7A	178.1 (2)	C2B—C1B—C6B—C7B	−178.6 (3)
C2A—C1A—C6A—C5A	2.3 (4)	C4B—C5B—C6B—C1B	−0.7 (4)
C2A—C1A—C6A—C7A	−176.4 (3)	C4B—C5B—C6B—C7B	178.4 (3)
C5A—C6A—C7A—O1A	167.5 (2)	C1B—C6B—C7B—O1B	−6.7 (4)
C1A—C6A—C7A—O1A	−13.9 (4)	C5B—C6B—C7B—O1B	174.1 (3)
C5A—C6A—C7A—C8A	−14.2 (3)	C1B—C6B—C7B—C8B	170.9 (3)
C1A—C6A—C7A—C8A	164.4 (2)	C5B—C6B—C7B—C8B	−8.2 (4)
C9A—O2A—C8A—C7A	171.1 (2)	C9B—O2B—C8B—C7B	75.9 (3)
O1A—C7A—C8A—O2A	4.4 (3)	O1B—C7B—C8B—O2B	−2.8 (4)
C6A—C7A—C8A—O2A	−173.96 (19)	C6B—C7B—C8B—O2B	179.4 (2)
C8A—O2A—C9A—O3A	1.6 (4)	C8B—O2B—C9B—O3B	2.2 (4)
C8A—O2A—C9A—C10A	−178.4 (2)	C8B—O2B—C9B—C10B	−179.9 (2)
O3A—C9A—C10A—C11A	16.9 (4)	O3B—C9B—C10B—C11B	−176.8 (3)
O2A—C9A—C10A—C11A	−163.1 (2)	O2B—C9B—C10B—C11B	5.3 (4)
O3A—C9A—C10A—C15A	−164.6 (3)	O3B—C9B—C10B—C15B	5.4 (4)
O2A—C9A—C10A—C15A	15.4 (3)	O2B—C9B—C10B—C15B	−172.5 (2)
C15A—C10A—C11A—F1A	−177.9 (2)	C15B—C10B—C11B—C12B	2.1 (4)
C9A—C10A—C11A—F1A	0.7 (4)	C9B—C10B—C11B—C12B	−175.8 (3)
C15A—C10A—C11A—C12A	1.1 (4)	C15B—C10B—C11B—F1B	−176.4 (3)
C9A—C10A—C11A—C12A	179.7 (2)	C9B—C10B—C11B—F1B	5.7 (4)
F1A—C11A—C12A—C13A	178.9 (3)	F1B—C11B—C12B—C13B	177.6 (3)
C10A—C11A—C12A—C13A	−0.2 (4)	C10B—C11B—C12B—C13B	−0.9 (5)
C11A—C12A—C13A—F2A	−179.8 (2)	C11B—C12B—C13B—F2B	−179.2 (3)
C11A—C12A—C13A—C14A	0.1 (4)	C11B—C12B—C13B—C14B	−0.5 (5)
F2A—C13A—C14A—C15A	178.8 (2)	F2B—C13B—C14B—C15B	179.2 (3)
C12A—C13A—C14A—C15A	−1.1 (4)	C12B—C13B—C14B—C15B	0.5 (5)
C13A—C14A—C15A—C10A	2.1 (4)	C13B—C14B—C15B—C10B	0.9 (5)
C11A—C10A—C15A—C14A	−2.1 (4)	C11B—C10B—C15B—C14B	−2.1 (4)

C9A—C10A—C15A—C14A	179.4 (2)	C9B—C10B—C15B—C14B	176.0 (3)

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
C8A—H8AB···O3B	0.97	2.60	3.451 (4)	147
C8A—H8AA···O1B^{i}	0.97	2.42	3.294 (3)	149
C5B—H5BA···O3A^{ii}	0.93	2.50	3.376 (4)	158
C8B—H8BB···O3A^{ii}	0.97	2.58	3.415 (3)	144
C14B—H14B···O1A^{iii}	0.93	2.59	3.216 (5)	125

Symmetry codes: (i) *x*, *y*−1, *z*; (ii) −*x*+1, *y*+1/2, −*z*+3/2; (iii) −*x*+1, −*y*+1, −*z*+2.

Fig. 1

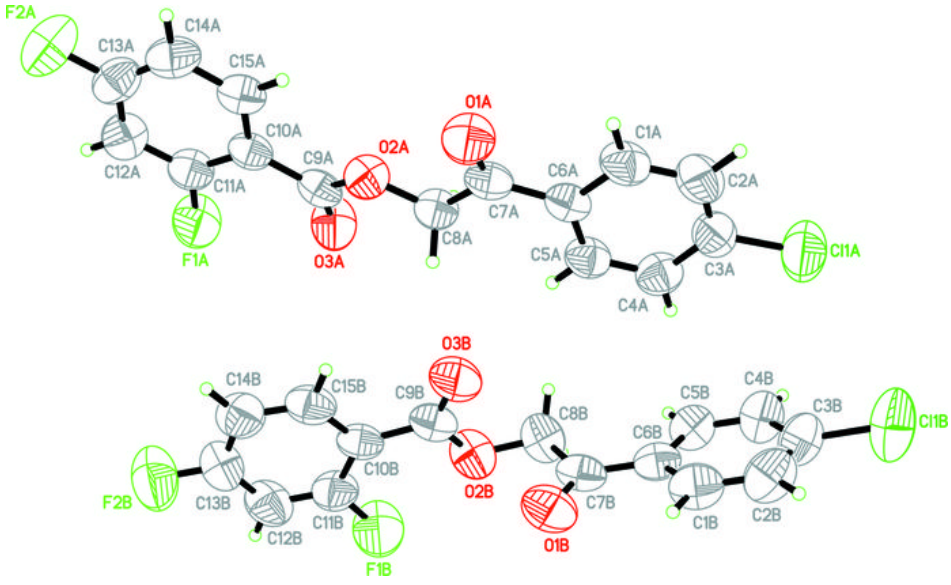
F2A
C14A
C13A
C15A
C10A
C9A
O2A
O1A
C1A
C12A
C11A
C6A
C2A
C7A
C8A
C5A
C3A
Cl1A
F1A
O3A
C4A
O3B
C14B
C15B
C9B
C8B
C5B
C4B
C3B
C6B
Cl1B
C10B
O2B
F2B
C13B
C7B
C11B
C1B
C2B
C12B
O1B
F1B

Fig. 2